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element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.  
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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \Rightarrow \text{H}_2\text{O}$ . 1. How many atoms are in a hydrogen molecule ( $\text{H}_2$ )? 2 2. How many atoms are in an oxygen molecule ( $\text{O}_2$ )? 2 3.

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equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the...

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This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that

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atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions.

## **Tenth grade Lesson Gizmo: Modeling Chemical Reactions**

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Worksheet Balance and classify five types of chemical reactions:

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synthesis,  
decomposition, single  
replacement, double  
replacement, and  
combustion. While  
balancing the  
reactions, the number  
of atoms on each side  
is presented as visual,  
histogram, and  
numerical data.

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